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Molecules can move from one physical state to another (phase change) and not change their atomic structure. Oxygen ( $O_2$ ) gas has the same chemical properties as liquid oxygen. The liquid state is colder and denser (less energy), but the molecules are the same.

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**Chem4Kids.com:  
Matter: Definition  
and Overview**

Terms in this set (...)  
matter. any substance  
possessing mass and  
volume; any substance  
possessing weight and  
volume. mass. the  
amount of matter  
composing of a  
substance. volume. the  
amount of space a  
substance occupies.  
weight.



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**Chemistry: Review  
of Matter  
Vocabulary  
Flashcards | Quizlet**

Chemical changes occur when bonds are broken and/or formed between molecules or atoms. This means that one substance with a certain set of properties (such as melting point, color, taste, etc) is turned into a different substance with different properties.

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Chemical changes are frequently harder to reverse than physical changes.

### **3.6: Changes in Matter: Physical and Chemical Changes ...**

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Test. PLAY. Match.

Gravity. Created by. ...

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and sucrose, or table sugar, are examples of molecules. Covalent bonds. A type of bond between atoms in which electrons are shared. An extensive property depends on. The amount of matter that is present ...  
Change of matter ...

**Chemistry Unit 1:  
Matter and Change  
Flashcards | Quizlet**

Chemistry is the study of matter. Matter is

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made up of extremely tiny particles called atoms and molecules. Atoms and molecules make up the three common states of matter on Earth—solids, liquids, and gases. The particles of a liquid are attracted to one another, are in motion, and are able to move past one another.

**Molecules Matter -  
Middle School**

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## **Chemistry**

- Matter cannot be created or destroyed by a chemical change -  
During any chemical reaction, the quantity of matter remains unchanged. The mass of the products is always equal to the mass of the reactants

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